



Fig. 1. ^1H -N.m.r. spectrum of *P. cepacia* O-specific polysaccharide.

TABLE I

^1H -N.M.R. PARAMETERS FOR THE *P. cepacia* POLYSACCHARIDE

Proton	Chemical shifts (p.p.m.)			Coupling constants (Hz)		
	Unit A	Unit B	Unit C	Unit A	Unit B	Unit C
H-1	5.27	5.20	4.98	$J_{1,2}$ 1.6	$J_{1,2}$ 1.6	$J_{1,2}$ 1
H-2	4.40	4.36	4.26	$J_{2,3}$ 2.5	$J_{2,3}$ 2.5	$J_{2,3}$ 2.4
H-3	4.20	4.16	3.88	$J_{3,4}$ 9.8	$J_{3,4}$ 9.8	$J_{3,4}$ 9.6
H-4	3.70	3.70	3.59	$J_{4,5}$ 9.8	n.d. ^a	n.d.
H-5	4.40	4.02	3.59	$J_{5,6}$ 6.1	$J_{5,6}$ 6.1	$J_{5,6}$ 6.1
H-6	1.44	1.48	1.48			

^aNot determined.

An almost complete interpretation of the ^{13}C -n.m.r. spectrum of the polysaccharide, using selective heteronuclear $^{13}\text{C}\{^1\text{H}\}$ double-resonance (Table III), confirmed the structure assigned and analysis of the effects of glycosidation indicated that units A–C had the same absolute configurations. The relatively low-field resonances for C-2 of unit C and C-3 of units A and B (78.8, 78.5, and 79.0 p.p.m., respectively) accord with their positions of substitution (*e.g.*, *cf.* the data in *ref.* 4). The relatively large $^1J_{\text{C,H}}$ values (169.2 and 170.2 Hz, respectively) for the

TABLE II

N.O.e. DATA FOR THE *P. cepacia* POLYSACCHARIDE

Irradiated proton	Observed proton and n.O.e. values ^a					
	Unit A		Unit B		Unit C	
H-1 of unit A	H-2 0.12		H-3 0.16	H-4 0.06		
H-1 of unit B			H-2 0.21		H-2 0.27	
H-1 of unit C	H-2 0.16	H-3 0.09			H-2 0.08	H-3 0.09 H-5 0.18

^aValues >0.05 are listed. Appearance of n.O.e.'s smaller than 0.05 was due to large correlation times and spin diffusion which are characteristic of polymers. They will be discussed elsewhere.

TABLE III

¹³C-NMR CHEMICAL SHIFTS (p.p.m.) FOR THE *P. cepacia* POLYSACCHARIDE

Unit	C-1	C-2	C-3	C-4	C-5	C-6
A	103.2	68.7	78.5	71.7	70.3	18.0 ^a
B	102.6	71.3	79.0	72.7	70.2	17.9 ^a
C	98.1	78.8	74.5	73.5 ^b	73.8 ^b	17.85 ^a

^{a,b}Assignments may be interchanged.

anomeric carbons of units A and B, determined from the gated-decoupling spectrum, showed these units to be α , whereas the small value (159.1 Hz) for C-1 of unit C proved it to be β (ref. 5).

The relatively small α -effect (+3.5 p.p.m.) for C-1 of unit C and the relatively large (−3.25 p.p.m.) β -effect for C-2 of unit A, determined by comparison with published data⁶, indicated that these (1→3)- β -linked units had identical absolute configurations⁷. Further, the large α -effect (+8.2 p.p.m.) for C-1 of unit A and the small β -effect (−0.65 p.p.m.) for C-2 of unit B proved that these (1→3)- α -linked units also had identical configurations⁷. In accordance with Klyne's rule⁸, the $[\alpha]_D$ value of +22° (water)¹ of the polysaccharide showed that rhamnose was D.

Thus, the structure assigned¹ to the *P. cepacia* IMV 3181 polysaccharide has been confirmed and it was shown that n.O.e.'s arising from irradiation of anomeric protons are helpful for determining the sites and configurations of glycosidic linkages as well as sequences of monosaccharides in polysaccharides. It is noteworthy that all n.O.e.'s observed for the polysaccharide of *P. cepacia*, as well as those⁹ for the polysaccharide of *Shigella flexneri* O-serogroup Y, had negative values.

EXPERIMENTAL

N.m.r. spectra were recorded with an AM-300 (Bruker) instrument for solutions in D₂O at 50° [internal acetone (δ_{H} 2.23) and methanol (δ_{C} 50.15)]. N.O.e. data were obtained by the TOE method¹⁰ and performed in the difference mode; the time constants used were D1 = 4 s, relaxation delay; D2 = 0.5 s, build-up n.O.e.

The O-specific polysaccharide of *P. cepacia* IMV 3181 was isolated as described earlier¹.

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